

# SYNTHESIS, ISOLATION, AND X-RAY STRUCTURAL CHARACTERIZATION OF TRIFLUOROMETHYLATED C<sub>90</sub> FULLERENES: C<sub>90</sub>(30)(CF<sub>3</sub>)<sub>18</sub> AND C<sub>90</sub>(35)(CF<sub>3</sub>)<sub>14</sub>

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Two CF<sub>3</sub> derivatives of C<sub>90</sub>, C<sub>90</sub>(30)(CF<sub>3</sub>)<sub>18</sub> and C<sub>90</sub>(35)(CF<sub>3</sub>)<sub>14</sub>, have been isolated via HPLC from the products of a high-temperature trifluoromethylation of a C<sub>76</sub>–C<sub>96</sub> fullerene mixture with CF<sub>3</sub>I. Their molecular structures were determined by single crystal X-ray crystallography using synchrotron radiation. The addition patterns of the new compounds are discussed in comparison with those of the corresponding chlorinated C<sub>90</sub>.

**Keywords:** Fullerenes, C<sub>90</sub>, Trifluoromethylation, HPLC, Structure elucidation.

## 1. Introduction

Compared to C<sub>60</sub> and C<sub>70</sub>, the investigation of higher fullerenes has been hampered by their relatively low abundance in fullerene soot and due to the existence of cage isomers [1]. Structural characterization of pristine higher fullerenes is typically accomplished by means of <sup>13</sup>C NMR spectroscopy, which provides information on molecular symmetries. However, identification of higher fullerenes by this conventional method is not unambiguous in many cases since several isomers may exhibit the same molecular symmetry [2,3]. An efficient alternative is chemical derivatization of higher fullerenes followed by isolation and structural characterization of the derivatives thus obtained, as exemplified by several examples for C<sub>76</sub>–C<sub>96</sub> [4–12].

C<sub>90</sub> belongs to the group of higher fullerenes with magic (6*n*) number of carbon atoms, which usually possess higher abundance and richer isomerism compared to their neighboring members [13]. For C<sub>90</sub>, there are 46 topologically possible isomers obeying the Isolated Pentagon Rule (IPR) [1]. Experimentally, <sup>13</sup>C NMR spectra of chromatographic C<sub>90</sub> fractions were interpreted as showing the existence of five isomers of C<sub>2v</sub>, C<sub>s</sub>, C<sub>2</sub>, C<sub>1</sub>, and C<sub>2</sub> symmetry [14,15]. A comparison between the experimental and theoretically predicted <sup>13</sup>C NMR shifts allowed the establishment of most probable carbon cages, 28 (C<sub>2</sub>), 30 (C<sub>1</sub>), 32 (C<sub>1</sub>), 35 (C<sub>s</sub>), 40 (C<sub>2</sub>), 45 (C<sub>2</sub>) and 46 (C<sub>2v</sub>), which are present in the C<sub>90</sub> fractions [16]. X-ray crystallographic study of co-crystals of C<sub>90</sub> from three HPLC fractions (obtained from arc-discharge of Sm<sub>2</sub>O<sub>3</sub>-doped graphite rods) with Ni<sup>II</sup>(OEP) (OEP - octaethylporphirin) resulted in the structural confirmation of three isomers of pristine C<sub>90</sub>, 1 (D<sub>5h</sub>), 30, and 32 [17,18].

A trifluoromethylated derivative of C<sub>90</sub>, C<sub>90</sub>(CF<sub>3</sub>)<sub>12</sub>, was suggested to contain C<sub>1</sub>-C<sub>90</sub>(32) cage on the basis of its <sup>19</sup>F NMR spectrum [19]. X-ray crystallographic investigation of C<sub>90</sub>Cl<sub>*n*</sub> (*n* = 22, 24, 28, and 32) unambiguously confirmed the C<sub>90</sub> cages nos. 28, 30, 32,

34 ( $C_s$ ), 35, and 46 and significantly contributed to the study of their reactivity towards inorganic chlorides [20,21].

Herein, we report the synthesis, HPLC isolation, and X-ray structure elucidation of trifluoromethylated derivatives of two  $C_{90}$  isomers,  $C_{90}(30)(CF_3)_{18}$  and  $C_{90}(35)(CF_3)_{14}$ . Addition patterns are discussed and compared with those of the corresponding chlorinated  $C_{90}$ .

## 2. Results

A starting higher fullerene mixture (35-45 mg; MER Corp.) contained  $C_{76}$ – $C_{96}$  fullerenes with the highest abundance of  $C_{84}$  and small admixtures of  $C_{60}$  and  $C_{70}$  [22]. The reaction with  $CF_3I$  was performed at 420 °C in a glass ampoule for 48 h, whereas the reaction at 560 °C (in a quartz ampoule) lasted only 1 h (see [23] for more details). In both cases, the trifluoromethylation products sublimed in the colder parts of the ampoules contained mixtures of fullerene( $CF_3$ ) $_{2n}$  derivatives with  $2n$  in the range of 12 – 20 according to MALDI TOF mass spectrometric analyses (fig. 1).

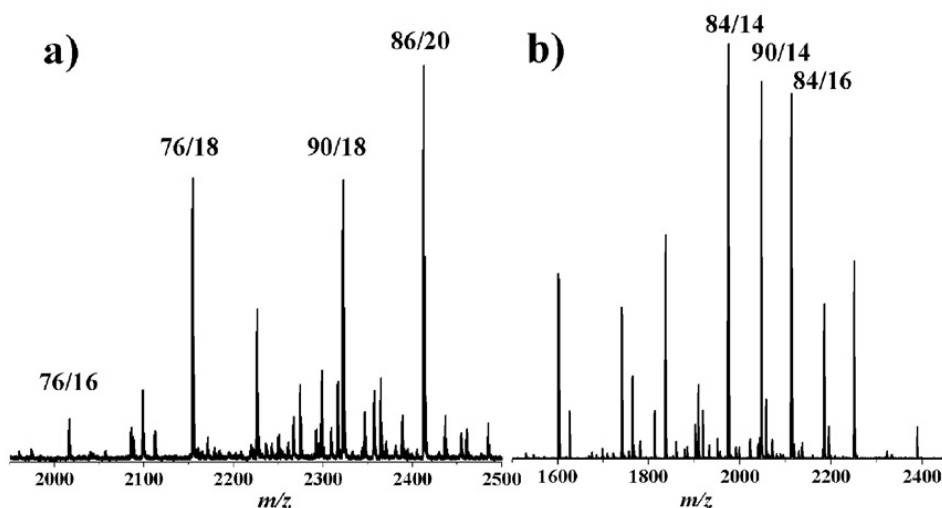


FIG. 1. Mass spectra of trifluoromethylation products. (a) reaction at 420 °C; (b) reaction at 560 °C. The compositions of  $C_{2m}(CF_3)_{2n}$  derivatives are given as  $2m/2n$ .

The product obtained at 420 °C was subjected to a two-step HPLC separation in hexane (Buckyprep column, 10 mm i.d.  $\times$  250 mm, flow rate 4.6 mL min $^{-1}$ , monitored at 290 nm). The fraction collected at 13.5 min in the second HPLC step contained mainly a  $C_{90}(CF_3)_{18}$  species. Slow concentration of the solution afforded small orange crystals of  $C_{90}(CF_3)_{18}\cdot 1.5$  Hexane, which have been investigated by single crystal X-ray diffraction using synchrotron radiation.

The sublimation product from the synthesis at 560 °C was first separated by HPLC in toluene at the same chromatographic conditions; the fraction eluted between 6.3 and 6.6 min was further separated using a toluene/hexane (1/1) mixture as the eluent. The fraction eluted at 19.4 min contained  $C_{90}(CF_3)_{14}$  species (fig. 2). Recrystallization from *o*-dichlorobenzene (*o*-DCB) afforded small red crystals. The crystal structure of  $C_{90}(CF_3)_{14}\cdot 2.5$  (*o*-DCB) was determined by X-ray single crystal crystallography using synchrotron radiation.

Synchrotron X-ray data were collected at 100 K at the BL14.2 at the BESSY storage ring (PSF at the Free University of Berlin, Germany) using a MAR225 CCD detector. Crystallographic data along with some details of data collection and structure refinements are

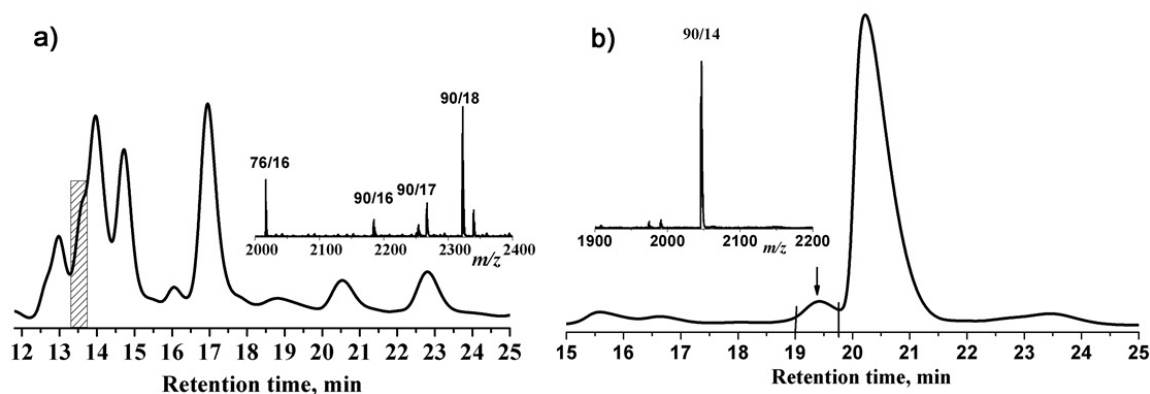


FIG. 2. Chromatographic isolation  $C_{90}(CF_3)_{18}$  (a) and  $C_{90}(CF_3)_{14}$  (b). The fractions collected are indicated by hatching (a) or arrow (b). Insets show mass spectra of the collected fractions. The compositions of  $C_{2m}(CF_3)_{2n}$  derivatives are given as  $2m/2n$ .

TABLE 1. Crystallographic data and some details of data collection and refinement for  $C_{90}(CF_3)_{18} \cdot 1.5\text{Hexane}$  and  $C_{90}(CF_3)_{14} \cdot 2.5(o\text{-DCB})$

Compound	$C_{90}(CF_3)_{18} \cdot 1.5\text{Hexane}$	$C_{90}(CF_3)_{14} \cdot 2.5(o\text{-DCB})$
$M_r$	2452.34	2427.41
Crystal system	Monoclinic	Triclinic
Space group	$C2/c$	$P\bar{1}$
$a$ [Å]	47.661(4)	14.206(1)
$b$ [Å]	22.404(2)	24.343(2)
$c$ [Å]	16.352(1)	25.359(2)
$\alpha$ [°]	90	93.312(3)
$\beta$ [°]	97.516(6)	99.786(8)
$\gamma$ [°]	90	102.104(3)
$V$ [Å <sup>3</sup> ]	17311(2)	8410.0(11)
$Z$	8	4
$D_c$ [g cm <sup>-3</sup> ]	1.882	1.907
Crystal size [mm]	0.02×0.02×0.01	0.05×0.02×0.01
Detector; $\lambda$ [Å]	MAR225 / 0.9050	MAR225 / 0.8856
Temperature [K]	100	100
$\theta(\text{max})$ [deg]	36.94	36.67
Refs collected / $R(\text{int})$	121922 / 0.051	122394 / 0.191
Data / parameters	17704 / 1712	33354 / 3099
$R_1$ [ $I \geq 2\sigma(I)$ ] / $wR_2$ (all)	0.060 / 0.160	0.090 / 0.216
$\Delta\rho$ (max / min) [e Å <sup>-3</sup> ]	1.107 / -1.265	0.391 / -0.426

presented in Table 1. The structures were solved with SHELXD and anisotropically refined with SHELXL. In the crystal structure of  $C_{90}(CF_3)_{18} \cdot 1.5\text{Hexane}$ , seven  $CF_3$  groups and solvated hexane molecules are strongly disordered. In the crystal structure of  $C_{90}(CF_3)_{14} \cdot 2.5(o\text{-C}_6\text{H}_4\text{Cl}_2)$ , there are two crystallographically independent  $C_{90}(CF_3)_{14}$  molecules. One  $CF_3$  group and three of six independent dichlorobenzene molecules are disordered. Crystallographic data are deposited under CCDC-917604 and -917603, respectively.

### 3. Discussion

Mass spectrometric MALDI analyses of trifluoromethylation products demonstrate the presence of  $C_{90}(CF_3)_{2n}$  species with  $2n$  ranging from 12 to 20, however, without information concerning the  $C_{90}$  cage connectivity and  $CF_3$  addition patterns. HPLC separation supported by subsequent MALDI MS analyses of separated fractions indicates the occurrence of several different  $C_{90}(CF_3)_{2n}$  isomers of the same composition whereas their assignment to specific  $C_{90}$  cages remained unknown. Growing crystals from separated fractions followed by X-ray crystallographic structure determination using synchrotron radiation was successful in only two cases of  $C_{90}(CF_3)_{14}$  and  $C_{90}(CF_3)_{18}$ , which are the first examples of unambiguous structural characterization of  $CF_3$  derivatives of  $C_{90}$  fullerene (fig. 3).

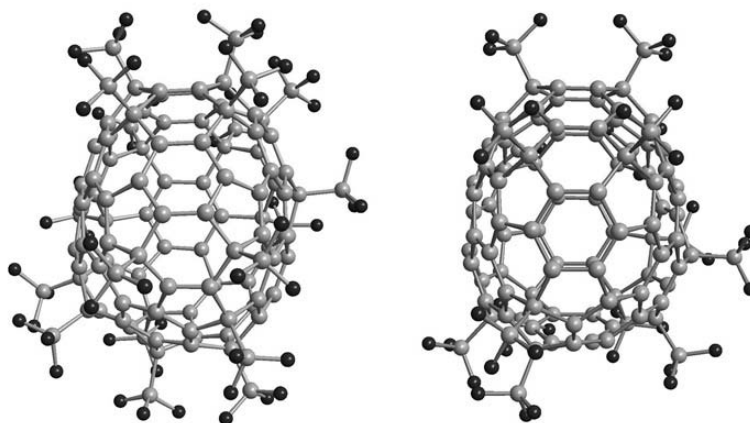


FIG. 3. Projections of the  $C_{90}(30)(CF_3)_{18}$  (left) and  $C_{90}(35)(CF_3)_{14}$  (right) molecules.

An analysis of the carbon cage connectivities in two molecules reveals that they contain different  $C_{90}$  cages,  $C_s\text{-}C_{90}(35)$  and  $C_1\text{-}C_{90}(30)$ , respectively, i.e., the  $C_{90}$  isomers, which have been already confirmed previously either in a pristine form (disordered  $C_{90}$  (30) cage in co-crystals with  $Ni^{II}(\text{OEP})$  [18]) or as chlorinated derivatives (both 30 and 35 isomers [21]). The description of molecular structures, their addition patterns, and comparison with corresponding chloro derivatives are convenient to perform using Schlegel diagrams (figs. 4 and 5).

In the  $C_1\text{-}C_{90}(30)(CF_3)_{18}$  molecule, 18  $CF_3$  groups are attached to a  $C_1\text{-}C_{90}$  cage exclusively in positions of double hexagon junctions (DHJs), whereas positions of triple hexagon junctions (THJs) remain unoccupied (fig. 4). All cage pentagons contain one or two attached  $CF_3$  groups. The stabilization of the addition pattern is achieved due to the formation of five isolated or partially isolated  $C=C$  bonds (av. bond length 1.33 Å) and one partially isolated benzenoid ring (av. C-C bond length 1.39 Å).

The previous X-ray crystallographic characterization of  $C_{90}(30)$  fullerene derivative has been performed on the crystal, which contained both  $C_{90}(30)Cl_{22}$  and  $C_{90}(28)Cl_{24}$  molecules in the unit cell [20,21]. A comparison of the addition patterns of  $C_{90}(30)(CF_3)_{18}$

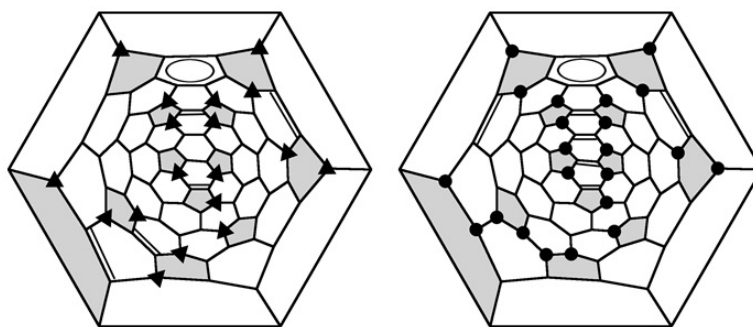


FIG. 4. Schlegel diagrams of  $C_{90}(30)(CF_3)_{18}$  (left) and  $C_{90}(30)Cl_{22}$  (right). Cage pentagons are highlighted with gray. Black triangles and circles denote attached  $CF_3$  groups and Cl atoms, respectively. The isolated or partially isolated C=C bonds and benzenoid rings are also indicated.

and  $C_{90}(30)Cl_{22}$  shows their rather close similarity because of 16 of 18 attachments positions in the former molecule are also occupied in the chlorinated molecule. This results in the same location of three C=C bonds and the benzenoid ring on the  $C_{90}(30)$  carbon cage. The differences in the addition patterns concern the attachment of some Cl atoms in adjacent (*ortho*) positions, which are less favourable for bulkier  $CF_3$  groups. For the same reason, the maximum number of  $CF_3$  groups attached to fullerene cages is only 20 (structurally confirmed for  $C_{70}$ ,  $C_{84}$ ,  $C_{88}$ , and  $C_{94}$ ) [23-26], whereas several fullerene chlorides with 32-34 Cl atoms are known [27,28].

The addition pattern of  $C_{90}(35)(CF_3)_{14}$  of both crystallographically independent (but chemically identical) molecules is characterized by  $CF_3$  attachment exclusively in positions of DHJs and the occupation of all twelve pentagons with  $CF_3$  groups, while two additional  $CF_3$  groups contribute to the formation of an isolated C=C bond (av. bond length 1.32 Å in two independent molecules) on the fullerene  $C_s$ - $C_{90}(35)$  cage (fig. 5). Three partially isolated benzenoid rings are also present on the cage (av. C-C bond length 1.40 Å).

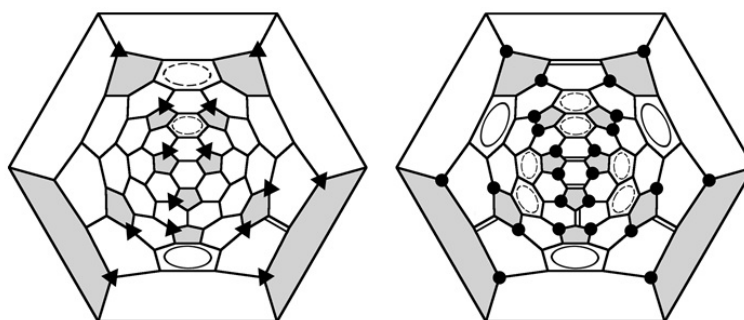


FIG. 5. Schlegel diagrams of  $C_{90}(35)(CF_3)_{14}$  (left) and  $C_{90}(35)Cl_{24}$  (right). Cage pentagons are highlighted with gray. Black triangles and circles denote attached  $CF_3$  groups and Cl atoms, respectively. The isolated or partially isolated C=C bonds and aromatic substructures are also indicated.

The first crystallographic confirmation of the cage connectivity of  $C_{90}(35)$  was reported for  $C_{90}(35)Cl_{24}$  and  $C_{90}(35)Cl_{28}$ , which possess rather similar chlorination patterns [20,21]. A comparison of the addition patterns in  $C_{1-}C_{90}(35)(CF_3)_{14}$  and  $C_s$ - $C_{90}(35)Cl_{24}$  demonstrates their similarity due to 12 common attachment positions. However, due to a large number of attached Cl atoms, several additions in *ortho* positions are present in the

$C_{90}(35)Cl_{24}$  molecule. Furthermore, all unoccupied carbon atoms on the  $C_{90}$  cage are involved into isolated aromatic substructures or isolated C=C bonds. Usually,  $CF_3$  and chloro derivatives of fullerenes show rather different addition patterns as can be exemplified by the comparison in pairs of  $C_{70}(CF_3)_{16}$  [29] and  $C_{70}Cl_{16}$  [30] or  $C_{76}(CF_3)_{18}$  [31] and  $C_{76}Cl_{18}$  [5].

#### 4. Conclusions

Trifluoromethylation of a higher fullerenes mixture followed by HPLC separation, crystallization, and X-ray crystallographic structure determination resulted in the first molecular structures of  $CF_3$  derivatives of  $C_{90}$  fullerene,  $C_{90}(30)(CF_3)_{18}$  and  $C_{90}(35)(CF_3)_{14}$ . The comparison of the addition patterns with those of the chlorides of the corresponding  $C_{90}$  isomers revealed their close similarities, whereas some differences can be attributed to different sizes of  $CF_3$  group and Cl atom as well as to different numbers of attached groups/atoms. It should be noted that  $C_{90}$  isomers 30 and 35 belong to the energetically rather stable isomers according to theoretical calculations [16,32,33]. However, the cage connectivity of the most stable  $C_2-C_{90}$  (45) still remains unconfirmed in any experimental report.

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