

Dielectric studies of nanocrystalline calcium tungstate

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PACS 78.67.Bf, 81.16.Be, 73.63.-b

DOI 10.17586/2220-8054-2016-7-4-599-603

Nanocrystalline samples of CaWO_4 were prepared at room temperature by simple chemical precipitation. The samples were characterized by X-ray diffraction and scanning electron microscopy. Energy dispersive X-ray analysis confirmed the elements present in the sample. The frequency and temperature dependence of the dielectric constant and ac electrical conductivity of the nanomaterial were investigated. Very low dielectric loss in nanocrystalline CaWO_4 powder was observed at high frequencies. The values of ac electrical conductivity calculated from the permittivity studies were found to increase as frequency increased, conforming to small polaron hopping.

Keywords: Chemical precipitation, dielectric constant, ac electrical conductivity, polaron hopping.

Received: 29 January 2016

Revised: 21 May 2016

1. Introduction

Nanocrystalline CaWO_4 has attracted particular interest because of its practical applications, such as laser host materials in quantum electronics and scintillators in medical devices [1–6]. It has been reported that CaWO_4 of scheelite-like structures is an excellent blue-emitting phosphor by their radiation of ultraviolet (UV) light [7]. Also, CaWO_4 has shown considerable promise as a fiber-matrix interlayer in oxide ceramic composites [8]. The lower dielectric constant and low loss make nanostructured CaWO_4 a promising candidate for applications as a low temperature co-fired ceramic (LTCC), substrate, and electronic packaging material [9]. Pullar et al. explained the microwave dielectric properties of AWO_4 ($A = \text{Mg, Zn, Ni}$ and Co) compounds with extrinsic parameter, such as density [10]. Sreedevi et al. reported that Ag_2WO_4 nanoparticles can be a promising material for the high dielectric constant gate in Si-based complementary metal oxide semiconducting devices [11]. The influence of processing methods on the characteristics of CdWO_4 powders and the related microwave dielectric properties were reported by Bao-Chun Guo et al. [12]. The study of dielectric properties of samples as a function of temperature and frequency may help in identifying their potential applications [13]. The characterization of dielectric behavior is very important not only to the theory of the polarization mechanism but also from an application point of view, where knowledge of the temperature and the frequency dependence of dielectric constant are very important. The relative dielectric constant of the material determines its ability to store electrostatic energy.

Dielectric studies of CaWO_4 nanoparticles are incomplete and need further investigation. In the present work, we synthesized CaWO_4 nanoparticles by chemical precipitation followed by calcination. The samples were then characterized by X-ray diffraction (XRD) and scanning electron microscopy (SEM). The frequency and temperature dependence of dielectric properties of sintered pellets made out of the products were then investigated.

2. Materials and methods

Calcium nitrate $\text{Ca}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$ (99.8 %, Sigma Aldrich) and sodium tungstate $\text{Na}_2\text{WO}_4 \cdot 2\text{H}_2\text{O}$ (99.9 %, Alfa Aesar) analytical grade reagents were used for the preparation of CaWO_4 nanocrystals. The samples were prepared by reacting aqueous solutions of calcium nitrate and sodium tungstate (0.1 M each) at room temperature. The precipitate formed was centrifuged, filtered, washed with distilled water a number of times, and dried in an oven to get fine powders of calcium tungstate. S_1 and S_2 are samples of nanocrystalline CaWO_4 were calcined at 650 and 750 °C, respectively. XRD studies of these samples were conducted using Bruker D8 Advance X-ray diffractometer ($\lambda = 1.5406 \text{ \AA}$) with $\text{CuK}\alpha$ radiation in 2θ range from 20 to 80 °. The morphological analysis of CaWO_4 nanoparticles was carried out with a scanning electron microscope JEOL MODEL JSM-6390LV, operating at 20 kV measurements.

The calcined powder sample was cold pressed in the form of cylindrical pellets of diameter 13 mm and thickness $d \sim 1.5$ mm by applying a pressure of ~ 10 GPa using a hand operated hydraulic press. The pellets were then sintered at 500 °C. The density of the pellet was determined to be 4.88 g/cm³. The circular faces of the pellets were made electrically conducting by coating with silver paste. Dielectric measurements as a function of frequency in the range of 100 Hz – 1 MHz were measured at various selected temperatures from 303 – 423 K using an LCR meter (Wayne Kerr H-6500 model) in conjunction with a portable furnace and temperature controller (± 1 K).

3. Results and discussion

The powder XRD spectra of CaWO₄ nanoparticle samples are shown in Fig. 1. Both the samples showed characteristic peaks of scheelite structure with tetragonal unit cell. The ‘d’ values taken from the JCPDS file No. 77-2235 for CaWO₄ are in close agreement with the observed ‘d’ values.

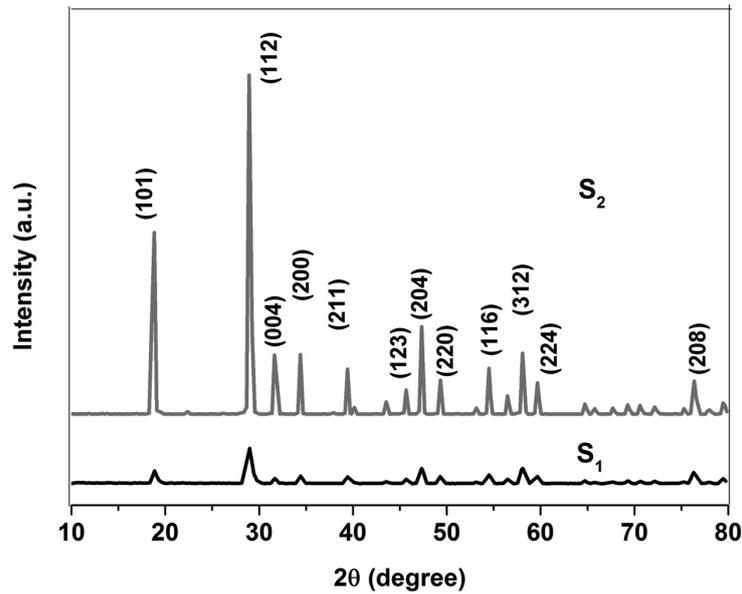


FIG. 1. XRD spectra of CaWO₄ samples

In general, the nanocrystallite size can be estimated from the Scherrer's formula: $D_{hkl} = K\lambda/(\beta \cos \theta)$, where λ is the x-ray wavelength (0.15405 nm), β the full-width at half maximum, θ the diffraction angle, K is a constant (0.89) and D_{hkl} the size along the (hkl) direction. From the analysis, the average crystallite size obtained was 39 nm for S₁ and 44 nm for S₂.

The SEM image of CaWO₄ nanoparticles calcined at 650 °C is shown in Fig. 2(a). They are clusters shaped like dumb-bells. The elemental analysis of the sample S₁ was done by energy dispersive X-ray (EDX) spectroscopy. Fig. 2(b) shows typical EDX spectrum of synthesized CaWO₄ nanoparticles. The peaks of the spectrum confirmed that the product contains Ca, W and O. The intense signal near at 1.774 keV indicates that W is the major element.

The dielectric constant and ac conductivity (σ_{ac}) were calculated by using equations $\epsilon' = Cd/\epsilon_0 A$ and $\sigma_{ac} = \epsilon' \epsilon_0 \omega \tan \delta$, respectively, where A is the face area, C the measured capacitance of the pellet, ϵ_0 the permittivity of vacuum, ω the angular frequency and $\tan \delta$ the loss tangent. Fig. 3(a) shows the variation of dielectric constant with frequency for temperatures from 303 to 423 K of sample S₁. It is seen that the dielectric constant for all temperatures are high at low frequencies which decreased rapidly as frequency increased, attaining a constant value at higher frequencies. For 303 K, the value of ϵ was 24.74 at 100 Hz, which decreased to 7.11 at 1 MHz. At 393 K, the values were 30.08 (100 Hz) and 7.15 (1.0 MHz). The corresponding values for 423 K were 39.54 at 100 Hz and 7.20 at 1.0 MHz. Fig. 3(b) shows a similar variation for samples S₁ and S₂, at 393 K. At lower frequencies the dielectric constant is found to be higher for the sample having smaller grain size (S₁), but approaches a constant value beyond 0.1 MHz. When temperature is increased, more and more dipoles are oriented, resulting in an increase in the dielectric constant for a given value of frequency [14]. At very high frequencies (MHz), the charge carriers would have started to move before the field reversal occurs and ϵ' falls to a small value at higher frequencies.

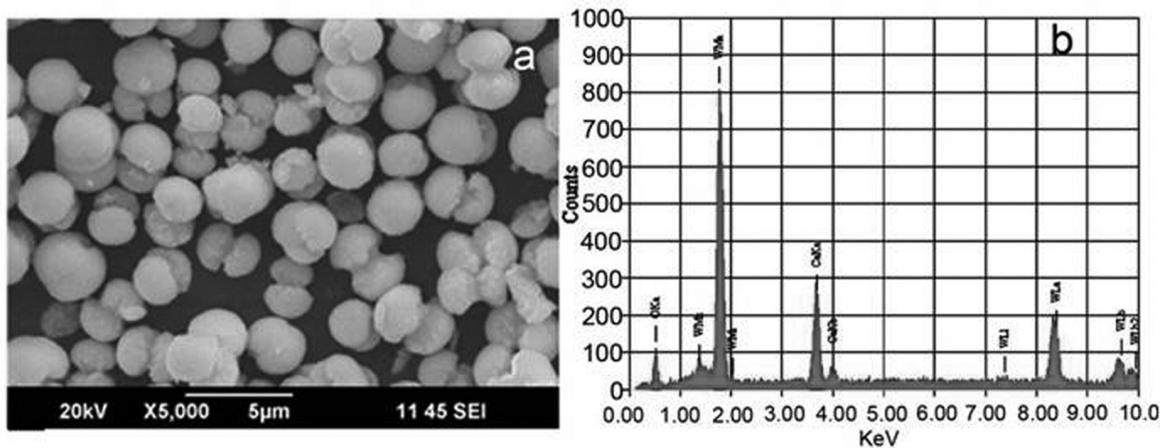


FIG. 2. (a) SEM image of CaWO₄ (S₁) and (b) EDX spectrum of CaWO₄

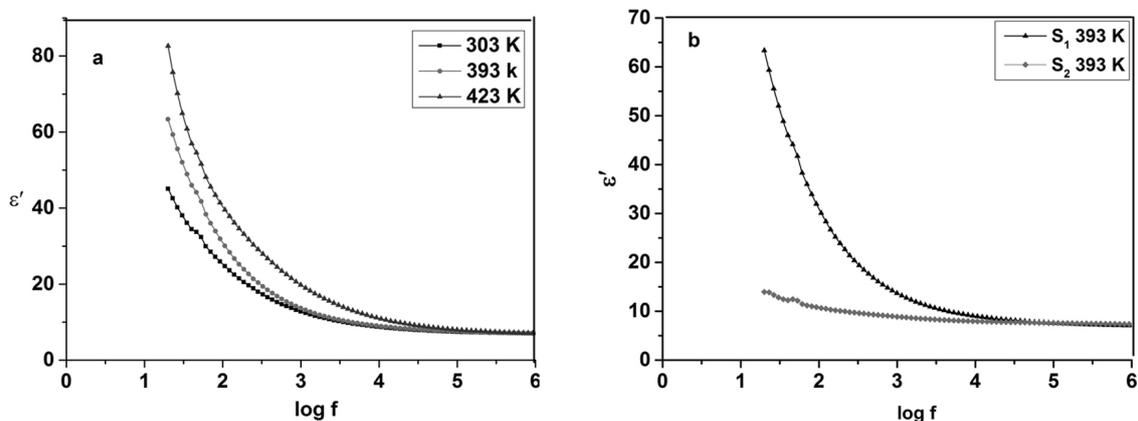


FIG. 3. The variation of dielectric constant with frequency of (a) sample S₁ at temperatures 303, 393 and 423 K and (b) samples S₁ and S₂ at 393 K

Space charge polarization and reversal of the polarization direction contributes much to the ϵ' [15]. With the increase in volume of the particle, the volume of the interfaces decreases. When volume increases, the contribution to ϵ' by electronic relaxation polarization inside the particles increases.

The frequency dependence of dielectric loss of sample S₁ is shown in Fig. 4(a). The loss factor represented by $\tan \delta$ has a value of 3.26 at 100 Hz which decreases slowly to zero at higher frequencies. At 393 K, the corresponding variation is not very different. For 423 K, $\tan \delta$ has a value of 9.53 at 100 Hz which gradually decreases almost to 0 at frequencies beyond 0.10 MHz. At 393 K the corresponding variation is not very different. The decrease in $\tan \delta$ takes place when the jumping rate of charge carriers lags behind the alternating electric field beyond a certain critical frequency. The inhomogeneities present in the interface layers in CaWO₄ nanocrystals produce an absorption current resulting in dielectric loss. This absorption current decreases with increase in frequency of the applied field. The hopping probability per unit time increases with increase in temperature. Correspondingly, the loss tangent also increases with increase of temperature [16]. The variation of $\tan \delta$ with frequency at 393 K for samples S₁ and S₂ with different grain sizes is shown in Fig. 4(b). At 100 Hz, the value is 3.4 for S₁ which decreases to 0.9 for S₂. This variation of $\tan \delta$ for different grain sizes is due to size effect [17]. The low value of $\tan \delta$ indicates its potential for microwave applications.

The loss in CaWO₄ can be explained using the electronic hopping model, which considers the frequency dependence of the localized charge carriers hopping in a random array. This model is applicable for materials in which the polarization responds rapidly to the appearance of an electron on any one site so that the transition may be said to occur effectively into the final state [17]. In the high frequency region $\tan \delta$ becomes almost zero because the electron exchange interaction (hopping) cannot follow the alternatives of the applied ac electric field beyond a critical frequency.

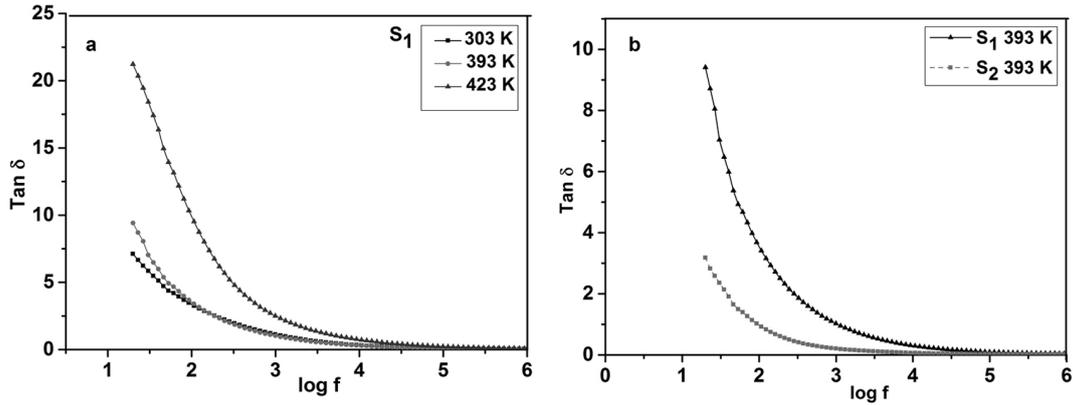


FIG. 4. The variation of loss tangent with frequency of (a) sample S₁ at temperatures 303, 393 and 423 K and (b) samples S₁ and S₂ at 393 K

Figure 5(a) shows the variation of *ac* conductivity (σ_{ac}) of sample S₁ with frequency. Initially, it has a small value which increased at higher frequencies. The nature of variation is similar for other temperatures, but the values are shifted upwards as the temperature is raised. For 303 K, σ_{ac} has a value of 5.001×10^{-7} S/m at 100 Hz which increased slowly at higher frequencies to 1.4×10^{-3} S/m at 0.10 MHz. At 393 K, the corresponding variation was not very different. For 423 K, the values were 2.236×10^{-6} S/m at 100 Hz and 2.781×10^{-5} S/m at 1.0 MHz. The variation of σ_{ac} with frequency at 393 K for different grain sizes is shown in Fig. 5(b). At 100 Hz, σ_{ac} is found to be 5.002×10^{-7} S/m for S₁ which increased to 6.88×10^{-5} S/m at 1.0 MHz. For S₂, the corresponding values were 5.823×10^{-8} S/m and 7.671×10^{-6} S/m. It is clear from the figure that the conductivity increased as frequency increased conforming to small polaron hopping [18]. Also, there is a possibility of conduction increased due to impurities at low temperature. It is found that at given temperature and frequency, σ_{ac} is higher for particle having smaller size. According to Elliot's barrier hopping model, *ac* conductivity increases with hopping distance [19]. Therefore, it may be concluded that in CaWO₄ hopping distance increased with reduction in particle size.

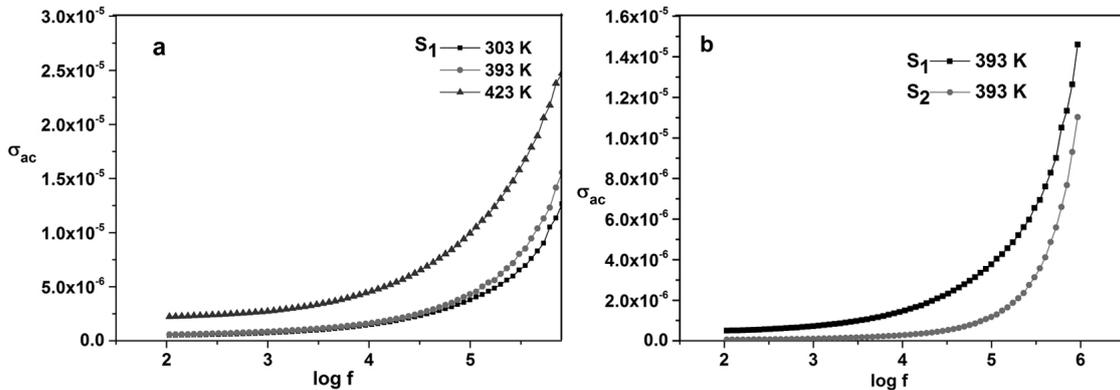


FIG. 5. The variation of *ac* electrical conductivity with frequency of (a) sample S₁ at temperatures 303, 393 and 423 K; (b) samples S₁ and S₂ at 393 K

4. Conclusion

The CaWO₄ nanoparticles were prepared at room temperature by simple chemical precipitation reaction without any catalyst, surfactant, or templates. The dielectric properties of CaWO₄ were determined as a function of frequency from 100 Hz to 1.0 MHz for temperatures ranging from 303 to 423 K. At lower frequencies, ϵ' and $\tan \delta$ have higher values while at higher frequencies the values reached steady lower values. Similar variation was observed when the temperature was raised but the values of ϵ' and $\tan \delta$ were elevated. The *ac* conductivity increased as frequency was increased conforming to small polaron hopping. The values of ϵ' , $\tan \delta$ and σ_{ac} showed considerable increase as the particle size was reduced. The very low value of loss tangent obtained for

CaWO₄ nanocrystals suggests that it is potentially useful for microwave applications. It was found that the applied frequency, temperature and particles size affect the dielectric properties of the CaWO₄ nanocrystals.

Acknowledgements

The authors are indebted to NSRC, Nirmala College, Muvattupuzha and Newman College, Thodupuzha for the support to undertake this study. The financial support from the University Grants Commission, New Delhi, India (FIP/12th Plan/KLMG020 TF03) is greatly acknowledged.

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